NITRATES IN INDIVIDUAL GROUNDWATER RESOURCES IN NITRA AND THEIR POSSIBLE RISKS TO THE POPULATION

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ARTICLE INFO

Received 20. 11. 2014
Revised 27. 11. 2014
Accepted 28. 11. 2014
Published 2. 2. 2015

OBJECTIVE

Until recently, the issue of groundwater was mainly focused on their use as a source of drinking water (about 75 % of European citizens are dependent on groundwater for drinking water), and as an important resource for industry (cooling purposes) and agriculture (irrigation). Groundwater is an irreplaceable part of the hydrological cycle and a key element in maintaining wetlands and flows in rivers during dry periods. It represents a basic drain (permanent year-round inflow to rivers) to surface hydrological systems, many of which are used as a source of water, or for recreational purposes (Lanz, 2001). The groundwater constitutes more than 50 % of the average annual flow during dry periods and the amount can exceed 90 % in many European rivers (Bodiš, 2010). The result is that the deterioration of groundwater quality may have direct effects on surface water and terrestrial ecosystems. Since groundwater flow is relatively slow, the effects of anthropogenic activities may be long-term (Tölgyessy et al., 2000).

The pollution caused by several decades ago as agriculture, industry, or other activities may threaten groundwater quality today, and in some cases even many generations in the future (Pado, 2001).

We can also say that the groundwater is "hidden resource", which is in quantitative terms more significant than surface water, while it is much more difficult to protect, monitor and eliminate pollution because of its inaccessibility. This "hidden" character is also the reason that the consequences of pollution are difficult to locate, describe and analyze (Bouchard, 2009; Mike, Shand, 2011).

Research shows that pollution from domestic, agricultural and industrial sources, despite of progress in some areas, still one of the main issues that impact on groundwater either directly through runoff processes or indirectly, through transfer of nitrate, whose content now exceeds Directive specified limit value in about one third of groundwater reserves in Europe (Bouchard, 2009). In European countries, the percentage of the population moves using the drinking water containing nitrate over 50 mg.dm\(^{-3}\) in the range 0.5 to 10 %, which is nearly 10 million people (Mike, Shand, 2011).

We use for drinking water in Slovakia underground water (82.2 %) and surface water (17.8 %). Largest natural reservoir of groundwater in Slovakia and Central Europe is Žitný ostrov with usable amount of about 20 400 Ls\(^{-1}\). In Bratislava, Trnava and Nitra are for a drinking water supply used only underground water sources. In other regions are for supply the population with drinking water used underground and surface water sources (Krč et al., 2007).

The city of Nitra (MŽP SR, 2012) is supplied from the water supply group of Nitra (districts Chrenová, Zobor, Staré Mesto - part Janíkovec) and water supply group Jelka - Galanta - Nitra (Staré mesto - part, urban parts Diely, Párovské Háje, Kynek, Mlynárce, Klokočina, Cermáň, Dolné a Horné Krškany).

Objective of this work was to gain knowledge about the quality of existing groundwater resources and demonstrably used for human consumption for drinking (adults or babies) in the administrative area of Nitra and Zobor, in terms of nitrate (NO\(_3\)) with reference to their potential health risks.

We chose for meeting the goals of groundwater resources: Svorad’s spring, Puškin’s spring, spring on Pivoňková street, spring on Mriańska dolina, spring Šindolka and spring Buganka) in the administrative area of Nitra and Zobor, also used for human consumption. The content of NO\(_3\) were assessed by Photocolorimetric method.

We also evaluate the results achieved in relation to the current legislation of the area. From the result of the performed analyzes during the whole period shows that the average concentration of NO\(_3\) represented in samples of water from the source Svorad’s spring 12.1 mg.dm\(^{-3}\), Puškin’s spring 14.6 mg.dm\(^{-3}\), from spring St. Martina 117.0 mg.dm\(^{-3}\), from spring on Pivoňkova street 6.4 mg.dm\(^{-3}\), from spring Šindolka 39.5 mg.dm\(^{-3}\) and spring Buganka 101.7 mg.dm\(^{-3}\).

The nitrate concentration exceeded the limit value in 16 % of cases in 2012 and it was 17 % of cases in 2013. Based on the measured values, therefore we do not recommend to use the water for human consumption from springs Buganka and St. Martina at the endpoint.

Keywords: Groundwater, quality, nitrates, city of Nitra

RESULTS AND DISCUSSION

We collected monthly and define the content of nitrate (NO\(_3\)) in groundwater samples from their individual sources in the period 2012 - 2013. The results were processed by the figure 1 and 2, which show that:

INTRODUCTION
Clinical manifestations of reduced oxygen transport

Bacterial reduction of nitrate (NO$_3^-$) to nitrites (NO$_2^-$) may take place in other parts of the digestive tract other than the stomach, where it occurs only at a reduced acidity (Paller et al., 2004). Nitrates have low toxicity and non-reducing environment and low concentrations are not for adults and for human beings harmful. However, in certain circumstances, it can be reduced to nitrates, which are toxicologically significantly more dangerous (Beresford, 1995).

Toxic effects (especially in children) are given mainly by reduction of nitrite and subsequent reaction of nitrite with hemoglobin. This occurs nitrate alimentary methemoglobinemia namely oxidation of Fe$^{2+}$ to Fe$^{3+}$ to convert hemoglobin (Hb) to a dark brown methemoglobin (MetHb), which is unable to carry oxygen (Darracq, Daubert, 2007). Clinical manifestations of reduced oxygen transport in the body mostly appears after exceeding 10 % MetHb concentration. This is most commonly observed in infants up to 3 months of age within 3% (Kratchovil et al., 2010).

The burden of population from drinking water containing chemical health risk is assessed on the basis of the highest detected concentration detected frequency of exceedances of limit values and comparing the tolerable daily intake. If the quality of drinking water in accordance with the requirements of Government Regulation no. 354/2006 Coll., then would not the lifelong intake of 2 liters a day should not be influenced negatively population health. In higher concentrations may affect the enzymes of the digestive system, absorption of certain nutrients, metabolism of vitamin A and thyroid function (EFSA, 2010).

The formation of nitrate alimentary methemoglobinemia shall apply in particular circumstances: children and adults, their dynamic physiological development, longer life expectancy, and thus longer exposure. Toxictox effects (especially in children) are given mainly by reduction of nitrite and subsequent reaction of nitrite with hemoglobin. This occurs nitrate alimentary methemoglobinemia namely oxidation of Fe$^{2+}$ to Fe$^{3+}$ to convert hemoglobin (Hb) to a dark brown methemoglobin (MetHb), which is unable to carry oxygen (Darracq, Daubert, 2007). Clinical manifestations of reduced oxygen transport in the body mostly appears after exceeding 10 % MetHb concentration. This is reflected in gray-blue discoloration of the skin around the mouth and ends of fingers and nose (cyanosis). When MetHb content is above 25 % it occurs weakness, increased heart rate and breathing, diarrhea, at 50-60 % MetHb it can lead to death. The normal concentration MetHb in man is within 2 %, in infants up to 3 months of age within 3% (Kratchovil et al., 2010).

Spring on Pivoňková street (φ = 48°02'73'', λ = 18°05'24''), is in a plantation valley (φ = 48°02'73'', λ = 18°05'24''), from this source melted water is maintained and roofing. It is the lowest altitude of 158 m asl from the forest, near the sanatorium under the hill Zbor (856.9 m asl). Seepage water using PE pipe with a diameter of 110 mm. During the year 2012, measured nitrate levels varied between 12.0 mg.dm$^{-3}$ (July) to 15.8 mg.dm$^{-3}$ (January). We found the minimum nitrate content of 6.7 mg.dm$^{-3}$ (March) to a maximum of 15.8 mg.dm$^{-3}$ (July) in 2013. It is generally believed that the content of NO$_3^-$ is in groundwater in natural conditions mainly controlled by the activity of microorganisms, that is, in particular nitrifying and denitrification processes. Precipitation water nitrate concentrations are low in Slovakia, averaging 2.65 mg.dm$^{-3}$ (Bodiš, 2010).

In soil water NO$_3^-$ concentration is generally increasing, mainly due to the biochemical transformation of ammonium (NH$_3^+$) present in the source (mainly rainfall) waters (Bodiš, 2010).

The measured nitrate concentrations in groundwater in 2012 compared with the limit value (Table 1, Figure 1) present in the sour

![Figure 1](image1.png)

**Figure 1** The measured nitrate concentrations in groundwater in 2012 compared with the limit value

Source sv. Martina is right at the road between the family homes in Martin’s valley (φ = 48°01’920”, λ = 18°00’8’2”), amounting to 194 m asl with the shelter built from bricks and with wooden roof. In terms of measured concentrations of NO$_3^-$ in 2012 has a minimum of 110.4 mg.dm$^{-3}$ in August, up to the extreme value of 129.8 mg.dm$^{-3}$ in March. We measured a minimum of nitrates in January, 2013 11.5 mg.dm$^{-3}$ to 132.4 mg.dm$^{-3}$ (January). It can be states in connection with such detected high levels of NO$_3^-$ in the samples that in terms of toxicological and health assessment is a particular risk group pediatric population, which is characterized by unique and different routes of exposure (Erkekoglu, Baydar, 2009).

In proportion to body weight receive a higher volume of fluid than older children and adults, their dynamic physiological development, longer life expectancy, and thus longer exposure.

![Figure 2](image2.png)

**Figure 2** The measured nitrate concentrations in groundwater in 2013 compared with the limit value

Toxic effects (especially in children) are given mainly by reduction of nitrite and subsequent reaction of nitrite with hemoglobin. This occurs nitrate alimentary methemoglobinemia namely oxidation of Fe$^{2+}$ to Fe$^{3+}$ to convert hemoglobin (Hb) to a dark brown methemoglobin (MetHb), which is unable to carry oxygen (Darracq, Daubert, 2007). Clinical manifestations of reduced oxygen transport in the body mostly appears after exceeding 10 % MetHb concentration. This is reflected in gray-blue discoloration of the skin around the mouth and ends of fingers and nose (cyanosis). When MetHb content is above 25 % it occurs weakness, increased heart rate and breathing, diarrhea, at 50-60 % MetHb it can lead to death. The normal concentration MetHb in man is within 2 %, in infants up to 3 months of age within 3% (Kratchovil et al., 2010).
The year and seepage water was using an iron pipe with a diameter of 50 mm. We found minimum content of NO₃ in February (33.5 mg dm⁻³) to 43.6 mg dm⁻³ in December in terms of the reporting period in 2012. We found that the nitrate content ranged from a minimum value of 35.1 mg dm⁻³ (January) to 44.3 mg dm⁻³ (June) in 2013. Spring Buganka (φ = 48019°S, λ = 18006°O), located on the street Lord's Valley, on private property at an altitude of 214 m asl, it is open to the public. We measured minimum content of nitrates during 2012 from samples we took in November (98.9 mg dm⁻³) to a maximum of 146.3 mg dm⁻³ in the month of January. The nitrate content range in June was 69.3 mg dm⁻³ to 95.8 mg dm⁻³ (January). All year above average levels of nitrates are risky not only for adults but especially for children. In general, the most sensitive part of the population are just infants under 3 months of age who are at risk when preparing baby food from water containing nitrate (Kožišek, 2007). They have several causes. One of those is a larger proportion of fetal hemoglobin, which is easily oxidized to the enzyme deficiency MetHb and MetHb - reductase which reduces back MetHb to Hb (Savino et al., 2006). Another cause may be higher pH in the stomach, allowing bacterial colonization and reduction of nitrate to nitrite. Formation of nitrites, and thus the risk of methemoglobinemia is increased in infections of the digestive tract, which are more common in children (Erkeloglu, Baydar, 2009). Another more sensitive population groups in the formation of MetHb are pregnant women and people with deficiency of glucose-6-phosphate dehydrogenase or MetHb reductase and people with reduced stomach acidity (achlorhydria, atrophic gastritis).

Determined the average nitrate levels found during the whole period 2012 ranged from samples from Svorad’s spring 13.2 mg dm⁻³, Puškin’s spring dm 12.7 mg dm⁻³, from spring St. Martina 120.4 mg dm⁻³, from spring on Pivónková street 4.6 mg dm⁻³, spring Šindolka 39.1 mg dm⁻³ and spring Buganka 124.2 mg dm⁻³. In addition to creating methemoglobinaemia, high nitrate concentrations, which reduced to nitrite, causing reaction with secondary and tertiary amines, which are present almost everywhere (vegetables, meat, dairy and grain products, eggs, beer and wine, medicines, pesticides, etc.) creation of nitrosamines (Murone et al., 2005). It has been shown that nitrosamines are able to induce tumor formation in all organs of the body, particularly the gastrointestinal tract, urinary bladder and lymphatic system, with the exception of bone tissue (Beresford, 1995).

The average nitrate content found in 2013 were in samples from water from Svorad’s spring 10.9 mg dm⁻³, Puškin’s spring 16.5 mg dm⁻³, from spring St. Martina 113.6 mg dm⁻³, from spring on Pivónková street 8.2 mg dm⁻³, spring Šindolka 39.9 mg dm⁻³ and spring Buganka 79.1 mg dm⁻³. Legal requirements for drinking water quality and its control constitutes § 11 of Act No.126 / 2006 Coll., setting out the mandatory provisions of the source in relation to the total content of nitrates, that enter the groundwater due to environmental pollution, agricultural activity, or they are created by nitrification activity of bacteria. The maximum allowed content of nitrates in leafy vegetables. EFSA Journal (2015)), 1–42.


REFERENCES


