

ENHANCED PRODUCTION OF SINGLE CELL PROTEIN FROM *M. capsulatus* (BATH) GROWING IN MIXED CULTURE

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doi: 10.15414/jmbfs.2016/17.6.3.894-899

ARTICLE INFO

Received 25. 2. 2016
 Revised 12. 8. 2016
 Accepted 7. 9. 2016
 Published 1. 12. 2016

Regular article



ABSTRACT

The growing global demand for nutritional protein means that a sustainable source such as Single Cell Protein from *Methylococcus capsulatus* (Bath) can become a potential replacement for fishmeal and other animal feeds. Improving biomass concentrations using statistical optimization during synergistic fermentations with a mixed consortium of the three heterotrophic bacteria *Alcaligenes acidovorans*, *Aneurinibacillus danicus*, and *Brevibacillus* sp. can increase the feasibility of the industrial process. The medium components Mg^{2+} , Ca^{2+} , Fe^{3+} , Cu^{2+} , PO_4^{2-} , NO_3^- , MoO_4^{2-} , trace metals, and process temperature were screened using a two-level Plackett-Burman Design in shake flasks which resulted in Cu^{2+} being the only significant factor. The optimum level of $CuSO_4 \cdot 5H_2O$ was found to be 40 μM using One Factor Response Surface Methodology, which was three times higher than the typical values of Cu^{2+} used previously. These combined strategies led to a 265% increase in biomass, with final cell concentration of 10.3 g/L, up from 2.8 g/L in fed batch fermentations over 48 hours. The heterotrophic bacteria did not grow on NMS or methane but increased biomass concentration when added to *M. capsulatus* (Bath) cultures.

Keywords: *Methylococcus capsulatus* (Bath), Heterotrophic bacteria, Single Cell Protein, Plackett-Burman, Response Surface Methodology

INTRODUCTION

Single Cell Protein (SCP) from *M. capsulatus* (Bath), which is cell biomass containing 70% protein, has been produced as a protein supplement for various animal feed applications. There is renewed interest in the industrial production of SCP since global food and protein demands are projected to rise significantly by 2050 in line with population growth (Alexandratos & Bruinsma, 2012; United Nations 2012). Its application as a solution to the impending world protein shortage depends on the development of the industrial bioprocess, which includes reactor design, efficient utilization of substrates, and medium development.

The fermentation of *M. capsulatus* (Bath) is conducted either as a monoculture or in a mixed culture with heterotrophic bacteria, and involves the use of Nitrogen Mineral Salts (NMS) medium and the input of air, natural gas, and carbon dioxide (Eccleston & Kelly, 1972; Stanley & Dalton 1982). A previous study (Harwood & Pirt, 1972) attempted to increase biomass concentrations from *M. capsulatus* (Bath) in a monoculture in bench scale experiments by investigating the effects of several ions, trace metals, and process parameters on biomass concentration. However, maximum cell concentration never increased beyond 0.4 g/L. Values for maximum cell concentration from other studies (Carlsen, Joergensen, & Degn, 1991; Furigo & Jorgensen, 1993; Larsen & Dalton, 1986; Leak & Dalton, 1986; Zhivotchenko, Nikonova, & Jorgensen, 1995) never exceeded a maximum of 6 g/L under various conditions. There has never been any attempt to optimize the medium for *M. capsulatus* (Bath) fermentations where heterotrophic bacteria are included.

The use of a mixed bacterial culture was found to be critical in the industrial production process used by Norferm Danmark A/S where three heterotrophic bacteria – DB 3 (*Alcaligenes acidovorans*), DB 4 (*Aneurinibacillus danicus*), and DB 5 (*Brevibacillus* sp.) – would “constantly invade” *M. capsulatus* cultures (Bothe et al., 2002). This study identified the heterotrophic strains, quantified the growth of the bacteria on substrates found in supernatant fractions, and determined that all strains were non-toxic. DB 3 and DB 4 exhibited high growth on acetate, DB 5 was best at reducing free amino acids, and all strains reduced total organic carbon. However, there was no indication from this study that these heterotrophs acted on the oxidation products of hydrocarbons other than ethane. Previous studies have indicated that the heterotrophic bacteria act in synergy with

an obligate methylophile like *M. capsulatus* (Bath) by scavenging the oxidized by-products of natural gas, as well as the metabolic by-products of the obligate methylophile, thereby bringing a measure of stability to the overall process while maximizing growth rates and yield coefficients (Harrison, 1978; Linton & Buckee, 1977). Thus it is important that the media for this process is optimized as a mixed culture since the presence of the heterotrophic bacteria is crucial for its industrial application.

In order to increase the volumetric productivity of SCP during continuous fermentations, there is a requirement for achieving high cell concentrations during the batch and fed-batch stages. The first optimization step must therefore involve improving the composition of the growth medium and optimizing other process parameters for achieving maximum cell concentrations in the shortest possible time. The simplest strategy for medium development is the ‘one at a time’ method where one component is varied while the others are kept constant. In this way, the individual effects of each component on the medium are determined, but it can be time-consuming due to the large number of experiments involved (Kennedy & Krouse, 1999). A better strategy for optimization is the Response Surface Methodology (RSM) where a functional relationship between the medium components and the response of interest can be determined (Khuri & Mukhopadhyay, 2010). It involves the use of a design matrix for screening along with an optimization technique in which a mathematical model is used to determine the optimized medium composition (Kennedy & Krouse, 1999).

Typically, a first-degree design (full or partial factorial) is used to screen and determine the significance of the factors. The most widely used screening method is the Plackett-Burman Design (PBD) which is a 2-level factorial design where the factor levels are set to a high and low value. In PBD, the number of experimental runs n is equal to $k + 1$, which is the same as the number of factors in the model, and the design only applies when n is a multiple of 4. This design does not account for interaction effects but if the levels are chosen correctly, it is a useful screening tool for determining the most significant factors, thereby reducing the number of factors to be optimized. The first-degree design is then followed by a second-degree design (single or multi-factor) which determines the optimum settings of each factor that result in either the maximum or minimum response over a certain range of interest. The most popular second-order design is Central Composite Design (CCD) which involves a 2^k factorial design with

factors set at five levels and the addition of an axial portion consisting of 2k axial points chosen based on the distance of the axial parameter (α) from the design center. In the case where the screening process results in only one factor being significant, a One Factor Response Surface Methodology (OFRSM) can be used to determine the optimal level of the variable.

OFRSM has been used to model methylene blue degradation (Attarchi, Montazer, & Toliyat, 2013), dynamic viscosity of a micro-emulsion (Jeirani et al., 2013), and to optimize biosurfactant production (Chen et al., 2012). These combined RSM techniques have been used for a wide range of applications including enhancing biomass from fungi (Zhang et al., 2014), microalgae (Zarate-Chaves et al., 2013), and yeast (Li et al., 2009). It has also been used to enhance ethanol production (Manwar, Mahadik, & Paradkar, 2013; Walia et al., 2014), and other high value products (Kosmider et al., 2012; Mokhtari-Hosseini et al., 2009; Yatsyshyn, Fedorovych, & Sibirny, 2010).

The objective of this study was to increase biomass concentration from *M. capsulatus* (Bath) in a mixed culture with three heterotrophic bacteria using statistical experimental design. This approach has not been previously used to optimize media for production of biomass by this microbe, either in a mono- or mixed culture. PBD was used as a screening tool for the medium components, followed by a second-degree RSM design chosen based on the resultant number of significant factors. Since only one factor was found to be significant, OFRSM was used to determine the optimal level. The predicted response determined from the OFRSM was then validated in shake flasks and fed-batch fermentations over 48 hours.

MATERIALS AND METHODS

Growth of *M. capsulatus* (Bath)

The methylotrophic microorganism used in this study was *Methylococcus capsulatus* (Bath) NCIMB 11132 (NCIMB Ltd. Culture Collection, Aberdeen, Scotland) obtained as freeze-dried vials. The cells were resuscitated using NCIMB Nitrogen Mineral Salts (NMS) Medium 131 (Salts per L: MgSO₄·7H₂O 1.0 g, CaCl₂·2H₂O 0.26 g, KNO₃ 1.0 g, C₁₀H₁₂FeN₂NaO₈ 0.0038 g, Na₂MoO₄·2H₂O 0.00026 g, Na₂HPO₄·2H₂O 0.356 g, KH₂PO₄ 0.026 g). Trace metals solution was added at final concentrations: CuSO₄·5H₂O 0.8 μM, FeSO₄·7H₂O 1.8 μM, ZnSO₄·7H₂O 1.4 μM, H₃BO₃ 0.24 μM, CoCl₂·6H₂O 0.21 μM, EDTA disodium salt 0.67 μM, MnCl₂·4H₂O 0.1 μM, and NiCl₂·6H₂O 0.04 μM. All chemicals were of ACS Analytical grade sourced from SIGMA ALDRICH CO (St. Louis, MO), MALLINCKRODT BAKER (New Jersey, USA), and FISHER SCIENTIFIC (New Jersey, USA). Cells were stored in 15% glycerol at -80°C and were also maintained on agar plates made from NMS medium with 0.02% v/v methanol added after sterilization. Cells were spread onto the plates aseptically using a vial from glycerol stock, after which the plates were placed in an airtight gas jar and left in an incubator set at 45°C, under an atmosphere of CH₄: Air: CO₂ of 9:9:2. Plates were then stored at 4°C until ready for use.

To prepare seed cultures for experiments, colonies from agar plates were inoculated into 500 mL baffled shake flasks containing 100 mL aliquots of NMS medium and 20 μL methanol. Each flask was sealed with a two-holed rubber stopper fitted with stainless steel tubing connected to silicone tubing with a sterile filter (Sartorius Midisart 2000) attached at one end to allow sparging with 45% CH₄ and 4.5% CO₂. Both pieces of tubing were clamped and the shake flasks were placed into a digital incubated shaker (ThermoScientific MaxQ Mini SHKE 4450 Digital) at 45°C and 180 rpm for 48 hours. Cells were centrifuged at 5000 g for 10 minutes, after which the cell pellet was re-suspended in NMS medium and used in further experiments. OD at 540 nm was measured using a ThermoScientific Genesys 6 UV/Visible Spectrophotometer with deionized water as the blank. Dry cell weight was found by vacuum filtering a known volume of cell culture through a Merck Millipore cellulose acetate 0.45 μM filter and dried to constant weight.

Growth of heterotrophic bacteria

For shake flask investigations, the three heterotrophic bacteria - NCIMB 13287 *Alcaligenes acidovorans*, NCIMB 13288 *Aneurinibacillus danicus*, and NCIMB 13289 *Brevibacillus* sp. were added to the culture. *A. danicus* and *Brevibacillus* sp. were cultured in 3 g/L Yeast extract broth (Lab M) and *A. acidovorans* was cultured in 3 g/L Nutrient broth (Lab M). 50 mL of each respective media was placed in a 250 mL shake flask which was fitted with a foam stopper and sterilized at 121°C for 15 minutes. Once cooled, the flasks were inoculated with

the respective organisms and placed in the shaker at 45°C, 180 rpm, for 24 hours. The cells were centrifuged, re-suspended in NMS before combining with *M. capsulatus* (Bath) for fermentation. For investigation of heterotrophic growth on NMS, each strain was added to 50 mL shake flasks containing 10 mL NMS to a start OD of 0.1. Flasks were sparged with air, methane, and carbon dioxide in the ratio described previously and incubated on the shaker and OD 540 nm was checked after 24 hours.

Plackett-Burman Design

Generation of the design matrix and statistical analysis for PBD was done using Minitab 16 (MINITAB Inc.). Media recipes used in previous literature (Table 1) were used to select the high (+) and low (-) levels for each variable. Nine (9) variables (Table 2) were selected as the most relevant to screen using a 12-run PBD (Table 3) for increasing biomass production in the fermentation broth. In the NCIMB medium, CuSO₄·5H₂O is only present in a small amount in the trace metal solution. However, it was included here as a major factor since it is included in other fermentation media for growth of *M. capsulatus* (Bath) in varying concentrations. The concentration and composition of the trace metals solution was kept constant but the volume added was modified. In the PBD model, all interactions were ignored and it was assumed that a first order regression model was applicable.

Each run in the design matrix was conducted in 250 mL shake flasks with 50 mL of medium corresponding to the specific run in the design matrix and inoculated with 1 mL of seed culture. Flasks were sparged and placed in an incubated shaker set at the appropriate temperature (either 37°C or 45°C) with a shaking speed of 180 rpm. After 24 hours, the OD₅₄₀ was measured. All experiments were carried out at least in triplicate and the results were the average of the replicate experiments with standard error ≤ 5%. The confidence level was set at 95% and the significant factors were selected based on the *p*-value falling below 0.05. The Effect (E) of each factor was determined from the regression analysis. The effects of the non-significant variables were used to set the new concentrations of these variables for validation experiments in shake flasks and fed-batch fermentations.

One Factor Response Surface Methodology

Based on the outcome of the PBD, OFRSM was used to determine the optimum concentration of the significant factor using Design Expert 8.0 (STAT-EASE Inc.). The factor was examined at five concentration levels in an eight run design generated by the statistical software to determine the optimum concentration. Shake flasks were prepared as previously described for PBD with the appropriate concentrations of the medium components.

Validation experiments

All experiments with the optimized NMS medium were performed in shake flasks in triplicate as previously described for 24 hours. The optimized medium was also tested via fed-batch fermentations in a 5L Sartorius B. Braun Biostat B-plus fermenter (SARTORIUS STEDIM). Start volume was 2L of the optimized NMS medium, with a ten-fold concentrated feed stream of the media at flow-rate 4 m L/min connected to low flow peristaltic pumps (Watson Marlow 101U/R) for fed-batch operation. pH was controlled automatically using 1M solutions of NaOH and H₂SO₄ and temperature was automatically controlled at 43°C in the double jacketed vessel. Agitation rates were selected based on achieving a mass transfer coefficient (kLa) value of 0.05 s⁻¹ for 0.5 vvm air in NMS medium (500 rpm), and 0.12 s⁻¹ (800 rpm) for higher cell densities (3-4 g/L). kLa was determined using the Hydrogen Peroxide-Catalase Method (Hickman, 1988; Cooke et al., 1991). Air flow rate was set at 0.5 vvm, CH₄ at 0.1 vvm, and CO₂ at 0.0012 vvm. *M. capsulatus* (Bath) cells and the heterotrophic bacteria were inoculated into the fermenter in the ratio 75:25 respectively, with *A. acidovorans* comprising 10% of the mixture, and 7.5% each of *Brevibacillus* sp. and *A. danicus*. All fermentations were stopped after 48 hours with final ODs recorded and compared to ODs at 48 hours for fed-batch fermentations using the un-optimized media. Results were converted to cell concentration (g/L) using a linear relation for OD versus dry cell weight (DCW) determined using cell suspensions of *M. capsulatus* (Bath):

$$OD = 2.6184 DCW + 2.0139 \quad (1)$$

Table 1 Media compositions for Nitrogen Mineral Salts medium from previous literature

Reference	(Foster & Davis, 1966)	(Whittenbury, Phillips, & Wilkinson 1977)	(Harwood & Pirt 1972)	(Dalton & Whittenbury, 1976)	(Carlsen, Joergensen, & Degn, 1991)	(Larsen & Dalton 1986)	(Furigo & Jorgensen, 1993)	(Zhivotchenko, Nikonova, & Jorgensen, 1995)
Methane %	50	30-50	50	20	2.4 g/L	Excess	40	66
Methanol v/v	-	0.01-0.1 (w/v)	-	-	-	-	-	-
Salt Concentration (g/L) unless otherwise stated							Concentration mM	
KNO ₃	-	1.0	-	1.0	1.0	-	NH ₄ ⁺ 25	20
NaNO ₃	2.0	-	-	-	-	-	SO ₄ ²⁻ 12.56	-
NH ₄ Cl	-	0.5	0.15	-	-	-	PO ₄ ³⁻ 7.81	-
NH ₄ NO ₃	-	-	-	-	-	1.88	Cl ⁻ 1.91	-
MgSO ₄ ·7H ₂ O	0.2	1.0	0.075	1.0	0.25	0.38	Mg ²⁺ 1.27	0.34
CaCl ₂ ·2H ₂ O	0.015	0.26	0.001	0.200	0.05	0.146	K ⁺ 1.86	-
^a NaFeEDDHA	-	0.004	-	-	-	-	Ca ²⁺ 0.94	0.59
^b NaFeEDTA	-	-	-	0.005	-	-	Na ⁺ 0.002	-
FeSO ₄ ·7H ₂ O	0.001	-	-	-	0.014	0.075	Fe ³⁺ 0.018	0.031
Na ₂ MoO ₄ ·2H ₂ O	-	0.00026	-	0.0002	0.001	0.0000012	Cu ²⁺ 0.0085	0.00037-0.128
MoO ₃	0.00001	-	-	-	-	-	Ni ²⁺ 0.00005	-
CuSO ₄ ·5H ₂ O	0.000005	0.00075	0.0003	-	0.0025	0.0035	Mn ²⁺ 0.00013	0.027
NaH ₂ PO ₄	0.09	-	-	-	-	-	Co ²⁺ 0.00053	0.00071
Na ₂ HPO ₄ ·12H ₂ O	0.21	0.1	1.2	0.33	-	0.356	Zn ²⁺ 0.0017	0.004
KH ₂ PO ₄	-	-	0.7	0.26	0.13	0.26	B ³⁺ 0.00026	0.08
Citric Acid	-	-	0.2	-	-	-	Mo ⁷⁺ 0.002	-
Ferric Citrate	-	-	0.002	-	-	-	Mo ⁶⁺ -	0.00083
HCl	-	-	0.003 ml	-	-	-	NaOH -	22.5-23.5
KCl	0.04	-	-	-	-	-	KH ₂ PO ₄ -	2.64
							NaH ₂ PO ₄ -	3.28

^aNaFeEDTA = Ethylenediaminetetraacetic acid ferric sodium salt C₁₀H₁₂N₂NaFeO₈

^bNaFe-EDDHA = "Sequestrene Iron complex" Sodium Ferric ethylenediamine BIS-(2-hydroxyphenylacetate) C₁₈H₁₆O₆N₂NaFe

Table 2 List of variables and their respective coded names for PBD screening

Variable	Variable code	Levels (g/L)		
		Low (-)	Center point (0)	High (+)
KH ₂ PO ₄	A	0.1	0.45	0.8
KNO ₃	B	1	1.5	2
MgSO ₄ ·7H ₂ O	C	0.5	1	1.5
CaCl ₂ ·2H ₂ O	D	0.1	0.3	0.5
C ₁₀ H ₁₂ FeN ₂ NaO ₈	E	0.005	0.05	0.1
CuSO ₄ ·5H ₂ O	F	0.003	0.011	0.018
Na ₂ MoO ₄ ·2H ₂ O	G	0	0.0005	0.05
Trace Metals ^a	H	0	1	2
Temperature ^b	I	37	41	45

^aTrace metals concentrations in mL/L

^bTemperature units = °Celsius

Table 3 Variables for PBD (in coded levels) with corresponding OD₅₄₀ response values for biomass concentration

Run #	Coded variables and levels											Response (OD ₅₄₀)
	A	B	C	D	E	F	G	H	I	*J	*K	
1	+	-	+	-	-	-	+	+	+	-	+	0.519
2	+	+	-	+	-	-	-	+	+	+	-	0.320
3	-	+	+	-	+	-	-	-	+	+	+	0.159
4	+	-	+	+	-	+	-	-	-	+	+	0
5	+	+	-	+	+	-	+	-	-	-	+	0.431
6	+	+	+	-	+	+	-	+	-	-	-	0
7	-	+	+	+	-	+	+	+	-	-	-	0.078
8	-	-	+	+	+	-	+	+	-	+	-	0.607
9	-	-	-	+	+	+	-	+	+	-	+	0.120
10	+	-	-	-	+	+	+	-	+	+	-	0.055
11	-	+	-	-	-	+	+	+	-	+	+	0
12	-	-	-	-	-	-	-	-	-	-	-	0.355

* J and K are dummy variables

RESULTS AND DISCUSSION

Screening of the factors by PBD

Nine variables were screened according to the PBD experimental runs shown in Table 3 along with the corresponding OD response values for each of the runs. The results show that runs 1, 5, and 8 gave the maximum yields while some runs showed no growth, or the combination of salts caused precipitation (represented as '0'). These results were then subjected to regression analysis (Table 4) where the effects of each factor are shown. Most effects were slightly positive, meaning that a higher level would produce a more favorable response while for CuSO₄·5H₂O, KNO₃, and temperature, the effects were negative indicating that a value lower than the maximum level would be more favorable.

Table 4 Regression analysis for results of PBD screening

Term	Effect	Coefficient	Standard error	t	P
Constant		0.2203	0.02398	9.19	0.012
KH ₂ PO ₄	0.0010	0.0005	0.02398	0.02	0.985
KNO ₃	-	-0.0557	0.02398	-2.32	0.146
MgSO ₄ ·7H ₂ O	0.0137	0.0068	0.02398	0.28	0.802
CaCl ₂ ·2H ₂ O	0.0780	0.0390	0.02398	1.63	0.245
C ₁₀ H ₁₂ FeN ₂ NaO ₈	0.0167	0.0083	0.02398	0.35	0.761
CuSO ₄ ·5H ₂ O	-	-0.1782	0.02398	-7.43	0.018
Na ₂ MoO ₄ ·2H ₂ O	0.1227	0.0613	0.02398	2.56	0.125
Trace Metals	0.0813	0.0407	0.02398	1.70	0.232
Temperature	-	-0.0118	0.02398	-0.49	0.671
S = 0.0830763	PRESS = 0.49692	R-Sq = 97.34%	R-Sq (pred) = 4.11 %		R-Sq (adj) = 85.35%

The results show that only $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ was significant with a p -value less than 0.05, which is confirmed in the half normal effects plot (Fig. 1) so the null hypothesis is rejected. $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ also had the largest coefficient and t -value (with effect in the negative direction). Even though the other eight variables were insignificant to the model with p -values above 0.05, the overall model is significant with a p -value of 0.012 and the R-Sq value showed that 97.34% of the total variation was explained by the model.

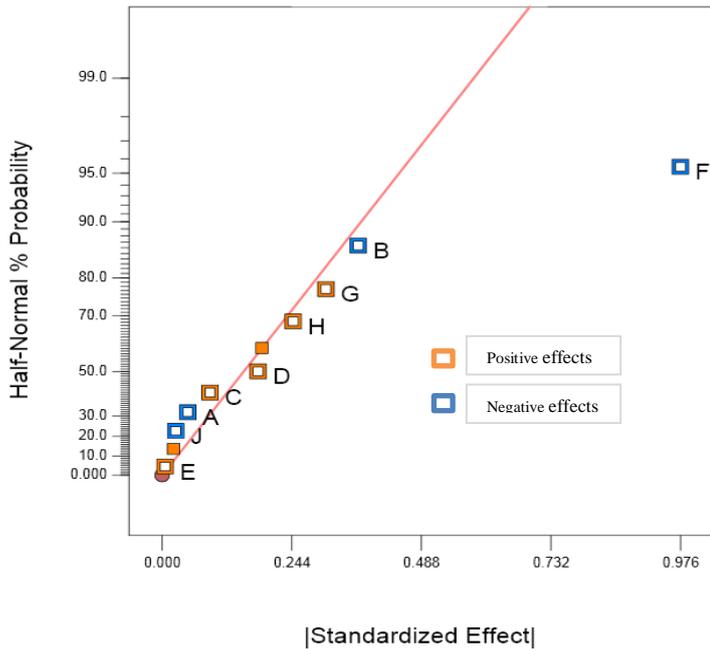


Figure 1 Half Normal plot of standardized effects showing significance of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (Variable F)

Optimization with One Factor Response Surface Methodology

Since only one variable was significant, the CCD could not be performed, hence OFRSM was used to determine optimal $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ concentration. The OFRSM run design consisted of eight runs at five concentration levels of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ in the range 0.003 – 0.018 g/L, with two replicates at the low, center, and high concentrations (Table 5). The concentration values for the other factors were set based on the direction and size of the effect values and were kept constant throughout the OFRSM optimization trials. The average ODs obtained from the OFRSM are also shown in Table 5. The highest ODs were observed around the midpoint value of the concentration range. As seen in the response surface map (Fig. 2) showing the distribution of the data between the limits of the confidence intervals (blue dotted lines). The prediction statistics generated from the software estimated that an OD of 0.804 would result when the concentration of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ is set at 0.0107 g/L. The confidence interval (CI) showed that 95% of the probable response values were contained within the interval 0.422 to 1.19.

Table 5 OFRSM design for $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ at five concentration levels with OD_{540} responses

Run	Concentration, (g/L)	Average OD_{540}
1	0.003	0.188
2	0.003	0.563
3	0.00675	0.106
4	0.0105	0.886
5	0.0105	0.717
6	0.0142	0.197
7	0.018	0.045
8	0.018	0.109

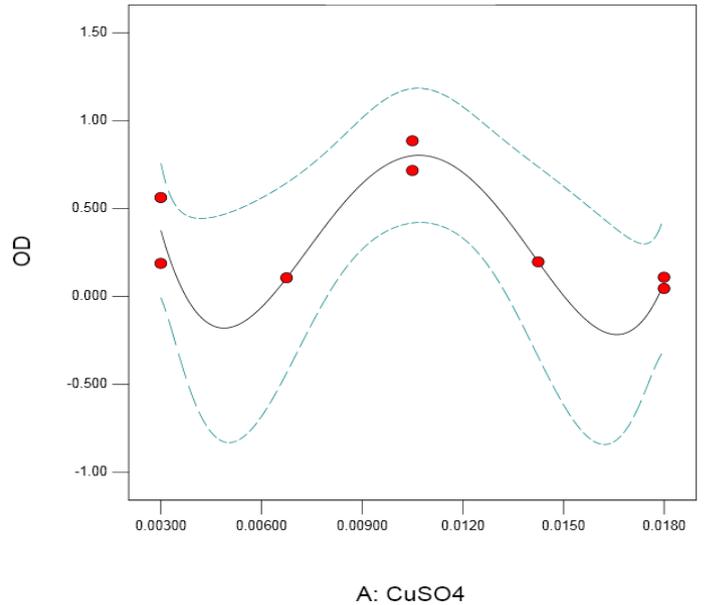


Figure 2 Response surface for $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ showing distribution of OD_{540} at the five design levels

Legend: - - - - Confidence intervals ——— Surface response plot

Confirmation experiments: Fed-batch fermentations

The shake flask experiments were performed using concentration of 0.01 g/L $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (40 μM) at a temperature of 43 °C. This resulted in an average OD of 0.935 ± 0.040 with a standard deviation (SD) of 0.057. This result was higher than the predicted value from OFRSM but was still within the range of both the confidence and prediction intervals, showing that the model was acceptable for forecasting the result with a certain degree of accuracy. The fed-batch fermentations over 48 hours showed an average OD of 29.0 or average cell concentration of 10.30 g/L (SD 0.24). With the un-optimized NMS medium, the average OD and cell concentration were 9.4 and 2.80 g/L (SD 0.18) respectively for the same time interval. This represented a 265% increase in cell concentration based on this new concentration level for $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$.

In the fed-batch fermentations, the feed stream was typically started between OD 4 - 7 when it was estimated that nutrients had become limited. In the optimized OPT run shown in Figure 3, the feed stream is started approximately 14 hours earlier than the UNOPT run, indicating a shortened lag phase due to the extra copper. The slope of the OPT run is also steeper, indicating a faster growth rate than for the UNOPT run. Another validation experiment in the fermenter was performed using the new concentration values of the other medium components with the initial un-optimized Cu^{2+} concentration value (0.00256 g/L). The resulting cell concentration was 5.60 g/L (SD 0.70), a 100% increase from the initial value of 2.80 g/L. This means that the overall increase in cell concentration was also affected by changes in values of the other factors. However, the medium composition was still limiting since the increase in copper allowed growth to further increase to 10.30 g/L for the same number of fermentation hours.

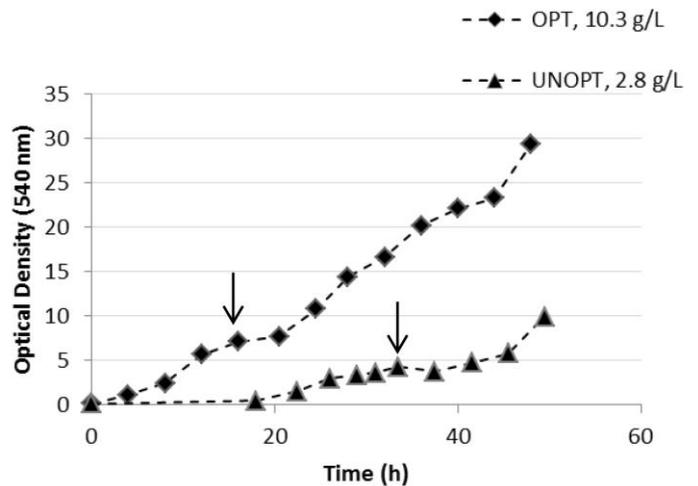


Figure 3 OD over 48 hours for fed-batch mixed culture
Legend: Before (UNOPT) and after (OPT) optimization. Arrows indicate where feed-streams were started.

Heterotrophic bacteria

All three bacteria demonstrated no growth on the NMS medium with CH₄ as carbon source but showed excellent growth in shake flasks in complex media when OD 540 nm was measured after 24 hours (*A. acidovorans*, 3.145; *A. danicus*, 1.823; *Brevibacillus* sp., 1.531). If the heterotrophs were CH₄ dependent, their growth would likely outpace that of *M. capsulatus* (Bath) since growth rates for the heterotrophs on acetate ranged from 0.45 h⁻¹ - 0.6 h⁻¹ (Stanley & Dalton, 1982) while the maximum growth rate recorded for *M. capsulatus* (Bath) in a monoculture was 0.37 h⁻¹ (Joergensen & Degn, 1987). Therefore the requirement for extra Cu²⁺ in the medium is not a requirement for heterotrophic growth but rather, it is necessary for optimal growth of *M. capsulatus* (Bath). When the three heterotrophs were combined with *M. capsulatus* (Bath), the OD increased from 0.349 ± 0.52 SD before addition to 0.807 ± 0.032 SD under the same conditions. The contribution of the heterotrophic cells to the overall OD in shake flasks was not determined.

It is well known from previous studies that the amount of copper in the medium influences the expression of either the soluble (sMMO) or particulate (pMMO) forms of Methane monooxygenase (Prior & Dalton, 1985; Semrau et al., 1995), with the pMMO being activated at Cu²⁺ concentrations greater than 1.5 μM (Zhitovchenko et al., 1995; Stanley et al., 1983). In the un-optimized medium used in this study, the Cu²⁺ concentration was 11 μM which indicates that the pMMO was already activated. However, the significant increase in growth when copper was increased to 40 μM in the optimized medium suggests that even though the pMMO was active, the Cu²⁺ concentration was still limiting. Zhitovchenko et al. (1995) indicated that 6 μM Cu²⁺ is the optimum level required for every 1 g/L biomass. Using this correlation for 10.30 g/L biomass concentration resulting from this study, the Cu²⁺ requirement would be approximately 62 μM (0.015 g/L). This concentration was investigated in the range of the OFRSM experiments in shake flasks but was not found to be optimal. This suggests that the relationship between biomass concentration and copper requirement is not a simple linear relation as the previous study suggests. From the data generated in this study, and assuming that 6 μM Cu²⁺ is required for 1 g/L biomass; the Cu²⁺ requirement can be modeled according to:

$$[Cu^{2+}] = e^{\left(\frac{X+6.72}{4.70}\right)} \quad (2)$$

The heterotrophic bacteria used for the mixed culture fermentations used previously by Norferm Danmark A/S was not planned but was found to be a necessary addition to the process. Their contribution to the cell population during batch culture was estimated by microscopic analysis to be approximately 5% of the cell population at an approximate cell concentration of 1 g/L. At Norferm Danmark A/S, the final heterotroph population, which was likely determined after extended continuous production runs, ranged between 13 % - 19.8 % of the total cell population, depending on the composition of natural gas used (Bothe et al., 2002). The activity of the heterotrophic bacteria as a function of dissolved oxygen concentration may help elucidate the function of these bacteria during different phases of the fermentation. Although the heterotrophs can function as natural gas oxidation product scavengers, their main purpose seems to be scavenging of dead cells since oxygen becomes limiting at higher cell concentrations resulting in less oxidation by-products of natural gas but increasing numbers of cells in death phase. The mixed culture growth is a complex process since the heterotrophs do not grow on NMS. Hence, modeling would have to include monitoring of production and consumption rates of all metabolic products during fermentation, natural gas oxidation products, and monitoring of cell population and viability, both as a consortium and individually.

Surprisingly, despite its crucial role in cell energetics, phosphate concentration was not a significant factor. Although calcium was also insignificant, the effect value dictated that a higher value was preferred so the concentration was increased to the mid-range value. Ca²⁺ is known to have an important role in the active sites of Methanol Dehydrogenase (MDH) and the MMO, allowing the prosthetic group pyrrolo-quinoline quinone (PQQ) to maintain the correct configuration for its role in substrate oxidation. The concentration of MoO₄²⁻ was doubled for similar reasons and this is understandable since molybdenum is an important element for biological processes. Overall, the statistical optimization allowed the system to be examined as a whole, and combined with the other changes to the media, this method worked well for detecting the optimum operating levels.

The impact of other fermentation parameters on cell density needs to be further investigated since the biomass concentration also depends on the availability of dissolved oxygen and methane. At low cell densities, these substrates are present in excess but as cell density increases, oxygen and methane become limiting. However, it is highly possible that the cultures in this study were grown under oxygen limitation since similar methanotrophic organisms have exhibited substrate inhibition as dissolved oxygen increased (Wendlandt, Jechorek, & Bruhl, 1993). The optimal agitation and gas supply rates need therefore to be investigated, and the point at which mass transfer becomes limiting needs to be determined for the overall optimization of this bioprocess.

CONCLUSION

The medium components for biomass production from a mixed culture of *M. capsulatus* (Bath) and three heterotrophic bacteria were screened by PBD and optimized with OFRSM. The use of these two experimental design techniques allowed for investigation of multiple process parameters in a condensed format. By using these tools, the optimum level of Cu²⁺ was found, but its effect at various concentrations could also be observed on the surface response map. The increase in growth rates and the almost tripled cell concentrations observed using the optimized medium shows the important role of copper in the operation of the methane oxidation enzyme systems. The complex nature of the heterotrophic synergy needs further investigation to fully understand their specific role in *M. capsulatus* (Bath) fermentations.

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