ENCAPSULATION OF POMEGRANATE SEED OIL USING W/O/W NANO-EMULSION TECHNIQUE FOLLOWED BY SPRAY DRYING AND ITS APPLICATION IN JELLY FORM.

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ABSTRACT
Punicic acid, which is an ω-5 polyunsaturated fatty acid representing the main component of pomegranate seed oil (PSO) with pharmaceutical and food usages, is susceptible to oxidation. Therefore, its water insoluble properties allow only few fractions to be absorbed. Therefore, the protection of this oil as well as bioavailability and good abortion properties are the aim of our work. Formation of water in oil in water (W/O/W) emulsion from pomegranate seed oil followed by spray-drying resulted in its transformation to nano-encapsulated oil powder whose nano-capsules had a narrow size distribution and small diameters in addition to zeta potential of about −22.9 mV. Thus, the oil-loaded system was assumed to be stable. Scanning electron microscope (SEM) measurement has confirmed the presence of surface pores which allow higher air permeability of powder, consequently a higher solubility and release was predicted. Addition of the obtained nano-encapsulated oil to jelly had a good sensory attribute and acceptability. In conclusion, pomegranate seed oil can be consumed as nano-capsules in jelly form with better water-soluble properties and it was predicted to have good absorption with expected great bioavailability.

Keywords: pomegranate seed oil, punicic acid ω-5, W/O/W nano emulsion, encapsulated oil, jelly form

INTRODUCTION

Fruit processing generates a huge amount of fruit waste and by-products annually worldwide which lead to serious industrial and environmental problems. So, maximizing the reuse of fruit waste and by-products is of great demand for food manufactures. In this respect, fruit by-products may be used as good sources of bioactive and nutritional ingredients for the production of functional foods. Pomegranate is considered one of the tropical fruits widely grown in many countries (Fadavi et al., 2005). Pomegranate juice containing high amount of phenolic compounds such as anthocyanin, which has strong antioxidant activities (Mahmoud et al., 2017), in addition it has the potential for exploration of broader applications as good candidate for functional foods and as nutraceutical plant-based products (Fouda et al., 2015). During the last few years, there are many pomegranate industrial products such as juice, jelly and jam that could produce great amounts of by-products like peels and seeds (Seeram et al., 2005). Seeds account for approximately 22% of the total wastes (rind plus seed) of pomegranate juice industries (FAO, 2011; Abd et al., 2017). Pomegranate seed is rich in oil that accounts for 20% of the seed weight and contains more than 95% triglycerides (Lansky and Newman, 2007). This oil contains punicic acid, approximately 80% (9c,11t,13c-18:3), linoleic acid, palmitoleic acid, palmitic acid, steric acid and arachidic acid (Hornung et al., 2002). Punicic acid is a conjugated linolenic acid (18:3 n-5) that belongs to the class of polyunsaturated fatty acids. Punicic acid has many health benefits due to its antioxidant and anti-inflammatory properties (Jing et al., 2012). Cytotoxicity for tumor cells by pomegranate seed oil was examined in both mouse and human monocytic leukemia cells (Suzuki et al., 2001) and it was found to be anticarcinogenic (Lansky and Newman, 2007). Also, it possesses strong antibesity, anti-diabetic, and antiproliferative effects as well as its significant effect on lipid metabolism (Aruna et al., 2016). These physiological effects make pomegranate seed oil of a great therapeutic potential, although more researches are needed to confirm the previous findings (De-Melo et al., 2014). Since punicic acid is highly susceptible to oxidation, addition of antioxidants is very important to increase its stability. But unfortunately, addition of antioxidant was found to decrease its antitumor activity. Also, the lake of solubility of punicic acid in water leads to absorption of only a minor fraction of it. Considering these previously mentioned characters, limited bioavailability of punicic acid is being a restricted factor that lowers its applications. Consequently, several trials were done to increase its applications including cyclodextrin inclusion, microencapsulation and mixing with edible oilseed fats. The application of encapsulation technology in functional food represents a novel area for researches. Nano-encapsulation of bioactive compounds could improve their dietary uptake, absorption and bio-availability of supplementary nutrients compared to ordinary sources (Mohammadi et al., 2016). Since, punicic acid is extremely susceptible to oxidation and insoluble in water. Therefore, the aim of this study is to increase solubility and, accordingly, to increase absorption and bio-availability as well as good protection of pomegranate seed oil by nano-encapsulation after the formation of double emulsion W/O/W and its application as a dietary supplement in the form of jelly.

MATERIALS AND METHODS
Preparation of pomegranate seed oil

Plant material: Pomegranate granatum fruit was purchased from local market, Cairo, Egypt at season 2018. Pomegranate seed oil was prepared as follows; after juice extraction, seeds were obtained then dried in air-oven at 40 °C. About 100 gm seeds were ground and extracted in duplicate with 2000 ml hexane at room temperature overnight. Then, the hexane was removed with a rotary evaporator at 40 °C under vacuum and the extract was dried to a constant weight then, it was stored at -18°C in the dark until being used.

Characterization of pomegranate oil fatty acid profile by GC-MS

The derivatization of fatty acids was preformed according to Rozes et al., (1993). Briefly, aliquot 20 mg of pomegranate oil was replaced into screw-capped tube and 1 ml of 1% sodium methoxide in methanol was added, then it was homogenized in a vortex for 20 sec. Then, 1 mL of hexane was added which led to separation of two layers, saponified (fatty acids) and unsaponified compounds; the organic layer (fatty acids) was collected, dried over anhydrous Na2SO4 and injected into GC-MS (Gas chromatography–mass spectrometry). The GC-MS system (Agilent Technologies) gas chromatograph (7890B) was equipped with mass spectrometer detector (5977A), Central Laboratories Network, National Research Centre, Cairo, Egypt. The GC was equipped with HP-5MS column (30 m x 0.25 mm internal diameter and 0.25 μm film thickness). Analysis was carried out using helium as the carrier gas at a flow rate of 1.0 ml/min, injection volume of 1 μl and
the following temperature program: 50 °C for 1 min; rising at 20 °C /min to 200 °C and held for 5 min; rising at 3 °C/min to 230 °C and held for 23 min. The injector and detector were held at 250 °C. Mass spectra were obtained by electron ionization (EI) at 70 eV and using a spectral range of m/z 20-550 and solvent delay 1.8 min. Identification of different constituents was determined by comparing the spectrum fragmentation pattern with those stored in Wiley and NIST Mass Spectral Library data.

**Nano-emulsion preparation**

W/O/W nano-emulsion was prepared by spontaneous emulsification according to (Assadpour et al., 2017) with some modification. Briefly, biopolymer solution was prepared; Arabic gum powder and maltodextrin (25 g each) were dissolved separately in boiling deionized water to prepare 200 ml solution. At the same time, aqueous solution of whey protein concentrate (50 g) was prepared by dispersing required amounts of whey protein (WPC) powder into 200 ml deionized water. Then, all prepared solutions were gently stirred for at least 30 min on a magnetic stirrer, then they were mixed together and stored overnight at room temperature for full hydration of biopolymers. The emulsion was prepared by mixing of biopolymer solution and Span 80 solution and then added oil as drop wise to the solution at high speed homogenization (20000 rpm) using a mechanical homogenizer (CAT, Unidrive 1000 D, M.Zipperer GmbH, Germany) followed by prop-ultrasonic (SONICS, Vibra-cell, VCX750, USA).

**Spray drying of W/O/W emulsion**

The prepared emulsion solution was transformed into encapsulated powder by spray drier (B-290, Buchi) equipped with a pressure air atomizing nozzle at 2.5 bar air pressure, inlet air temperature of 180 ± 5 °C, and outlet air temperature of 90 ± 5 °C with a feed flow rate of 450 ml/h. The dried powder was collected and stored in dark bottle, air tight containers at 4°C until further analysis.

**Encapsulation efficiency measurement**

The encapsulation efficiency was determined by calculating the difference between the amount of oil at the surface against encapsulated oil in capsules as described by Calvo et al., (2010). Briefly, 5 g of microcapsules were washed with 50 ml hexane under magnetic stirrer at 200 rpm for 60 sec. After 10 min, the mixture was filtered through filter paper (No. 41, Whatman, Maidstone, UK). The powder residue was again washed twice with 5 ml of hexane. Hexane was evaporated to constant weight by oven at 105°C. The total oil content was extracted by Soxhlet extraction of microcapsules using 250 ml hexane for 5 g of powder for 4 h. The encapsulation efficiency was calculated by the following equation:

Encapsulation efficiency (%) = (Total oil content - Surface oil content) / (Total oil content) ×100.

**Droplet size measurement**

Droplet size (particle size and zeta potential) of pomegranate nano-oil was analyzed using a dynamic light scattering method (Zeta-sizer Nano Zs, Malvern Instrument, Malvern, UK). To avoid multiple scattering, all samples were diluted by water. For microstructure analysis, a Zeiss optical microscope (Germany) was used and the obtained images were analyzed using Image J Software (Hosseini et al., 2015).

**Scanning electron microscopy of encapsulated powders**

The nano-encapsulated powder was sprinkled onto a two-sided adhesive tape and then coated with a thin layer of gold. Morphological features of particles were then observed by a field emission scanning electron microscope (S- 4160 Cold Field-Emission SEM, QUANTA, FEG 250, Thermo Fisher Scientific, USA) with an accelerated voltage of 320 kV and photographed at 6000.

**Fourier transform infrared spectroscopy (FTIR)**

The structure analysis of the samples was examined by Fourier transform infrared spectroscopy (FT-IR). The IR spectra of the samples were recorded by FT-IR 6000 spectrometer (JASCO, Japan). The spectrum was scanned in transmission mode from 400 to 4000 cm\(^{-1}\) wavenumber. The dry samples were blended with KBr powder and pressed into a disk before spectrum acquisition.

**Application in jelly form**

Jelly powder was prepared according to (Mahmoud et al., 2017) with some modification. Samples were prepared with added encapsulated oil at concentrations of 2.5, 5, 7.5 and 10%, while control sample was a commercial jelly powder. The treatments were first prepared by mixing 20 gm of gelatin powder and 75 gm of sugar with encapsulated oil, 0.1 gm of colors and 1 gm of strawberry essence powder. Next, 200 ml hot water was added to the powder in a mixing bowl and the solution was adjusted with 20% citric acid solution to pH of 4.3 at which point the preservative sodium benzoate (0.05%) was added. Prepared jelly with different concentrations of the added encapsulated oil was stored in a 4 °C refrigerator and sensory evaluated to determine the more preferred level of added concentration.

**Sensory evaluation**

Sensory evaluation was conducted in the Food Technology Department, National Research Centre, Egypt. Twenty trained panelists aged between 25 and 40 years were selected to take part in the sensory panel. The panelists measured the selected critical jelly attributes such as color, flavor, texture, appearance, after taste and overall acceptability according to Mahmoud et al., (2017). All evaluations were performed in a well-controlled laboratory. To determine the degree of like for jelly products (9 = like extremely, 5 = neither like nor dislike, 1 = dislike extremely) according to Peryam and Girardot (1952).

**Statistical analysis**

The data obtained from the study was statistically subjected to analysis of variance (ANOVA) Snedecor and Cochran (1980). Data were presented as mean ± standard deviation (SD). The least significant difference (LSD) value was used to determine significant differences between means using ASSIST version 7.7 beta. Values were considered significant at p ≤ 0.05.
RESULTS AND DISCUSSIONS

Pomegranate seed oil Fatty acid profile by GC-Mass spectrum

Pomegranate seed is a rich source of oil with special characteristics, because it contains high percentage of punicic acid which is an omega-5 fatty acid. Results of the present study revealed that total lipids of the Egyptian P. granatum seed was 15.38% on dry weight basis. Previous studies have indicated approximately similar results. Fadavi et al., (2006) reported that lipid content in sour pomegranate was 14.83% while Melgarejo and Artes (2000) reported that the lipid content was 6.3–12.2% for sweet Spanish pomegranate. Table 1 shows the fatty acid composition of oil extracted from pomegranate seed. It can be noticed that unsaturated fatty acids represent 89.73 % in pomegranate seed oil, where punicic acid (o-5) is the predominant fatty acid (64.91%) and punicic acid isomers named alfa and beta eleostearic acids were represented as 11.3 and 2.86, respectively. Palmitic acid is the major saturated fatty acid with low percentage (3.66 %). Dadashi et al., (2013) reported that punicic acid is the main fatty acid identified by gas chromatography for the oil extracted from four Iranian commercial varieties of pomegranate seed and it was found in the range from 72.07 % to 73.31 %. The difference in punicic acid percentage between our results and the results of Dadashi et al., (2013) may be attributed to pomegranate different varieties as well as different conditions of cultivation.

Table 1 Fatty acid composition of pomegranate seed oil by GC-Mass

<table>
<thead>
<tr>
<th>Fatty Acid</th>
<th>Relative %</th>
<th>Library</th>
</tr>
</thead>
<tbody>
<tr>
<td>Palmitic acid Methyl ester</td>
<td>3.66</td>
<td>NIST14.L</td>
</tr>
<tr>
<td>cis-Linoleic acid methyl ester</td>
<td>3.89</td>
<td>NIST14.L</td>
</tr>
<tr>
<td>Oleic acid methyl ester</td>
<td>5.91</td>
<td>W9N11.L</td>
</tr>
<tr>
<td>Stearic acid methyl ester</td>
<td>3.86</td>
<td>NIST14.L</td>
</tr>
<tr>
<td>Punic acid methyl ester</td>
<td>64.91</td>
<td>W9N11.L</td>
</tr>
<tr>
<td>Alfa-Eleostearic acid methyl ester</td>
<td>11.3</td>
<td>W9N11.L</td>
</tr>
<tr>
<td>Beta-Eleostearic acid methyl ester</td>
<td>2.86</td>
<td>W9N11.L</td>
</tr>
<tr>
<td>11-Eicosenoic acid methyl ester</td>
<td>0.86</td>
<td>NIST14.L</td>
</tr>
<tr>
<td>Oleic acid amide</td>
<td>2.72</td>
<td>NIST14.L</td>
</tr>
<tr>
<td>CLA Unsaturated fatty acid</td>
<td>79.07</td>
<td>--</td>
</tr>
<tr>
<td>Saturated fatty acid</td>
<td>7.52</td>
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</tr>
</tbody>
</table>

Water/Oil/Water emulsion

Emulsions are one of the promising encapsulation and delivery technologies; they have advantages such as controlled release and chemical stability of encapsulated nutrients like minerals, vitamins, amino acids, polyphenols, bioactive compounds etc. (Ilyasoglu and El, 2019; Prichapan and Klinkesorn, 2014). Double emulsions have many food applications such as reducing fat content, improving fatty acid profile, encapsulation of functional components and reducing sugar/salt content of foods by increasing taste intensity. Water-oil-water emulsions are primarily used in food applications. In this study, formation of double emulsion W/O/W by getting the oil particles wrapping the WPC molecules was made, then it was covered by another layer of polymer to produce a bigger molecule. That emulsion was obtained by either WPC and span twenty to produce water in oil in water emulsion. This structure allows more protection of oil, as well as its transformation into more water soluble particles and allow addition of more water-soluble additives to the emulsion (Buyukkestelli, and El, 2019).

Characterization of formulated Nano-capsules

The role of polyunsaturated fatty acids and their delivery to the intestine in sufficient amount is beneficial due to its anti-inflammatory properties and accordingly reduce the chronic diseases (Ilyasoglu and El, 2014). Therefore, characterization of the prepared biodegradable nanoparticles of pomegranate seed oil was done to predict the improvement in its stability and solubility.

Droplet size measurement

The Particle size and its distribution were measured using DLS (dynamic light scattering) method, which is the common and popular technique for determining nano droplet size distribution profile of suspension (Khazaei et al., 2014). The droplet size of the tested samples Figure 1 showed that oil-loaded polymer micelles had a narrow size distribution and small diameters (about 112 nm). That small size caused by adsorption onto the surface of a system or interface surfactant reduces the surface free energy and consequently decreases interfacial tension between the lipid matrix and the aqueous phase and leads to easier disruption of particles during homogenization to form smaller particles (Soleimanian et al., 2018). Zeta potential reflects the stability of emulsion systems (Soleimanian et al., 2018). Also, Zeta potential reflects the repulsion strength between charged particles. Typically, achieving a good stability required a minimum zeta potential of ±30 mV for nano suspension electrostatically stabilized. Furthermore, ±20 mV zeta potential is sufficient in the case of the combination between electrostatic and steric forces. This is because the hydrophilic surfactant coat can improve the stability of emulsion by further hydration surface layer (Tammjidi et al., 2014). In this study, the polymer displayed a zeta potential of about −22.9 mV, hence the oil-loaded polymer system was assumed to be stable. The oil-loaded polymer micelles had a great surface area because of the entrapment of polymer molecules, which may be accountable for the enhancement of the negative charge on the micelle’s surfaces. It was reported that by relatively high surface charges, particles could reinforce each other with a strong electrostatic emulsion force, thus improved the stability of the system (Chen et al., 2012).
Figure 1 Zeta Size and Potential of nano encapsulated pomegranate seed oil.

Encapsulated powder morphology by Scanning electron microscopy and Encapsulation efficiency

Figure 2 reveals the SEM microphotographs of the powder containing encapsulated pomegranate seed oil with high content of ω-5 punicic fatty acid. Encapsulated powder consisted of WPC particles in the size of nano (about 112 nm) wrapped with pomegranate seed oil layer and every group (about 2-4 molecules) of nano particles covered by another layer of polymer to produce a bigger particle (ranged from 200 up to 2000 nm) as shown in Figure 2. Observations obtained from SEM confirmed the presence of surface pores obtained by emulsion matrix containing Span as a surfactant. This structure allowed great air permeability into the powder, while at the same time, much solubility and release are predictable due to higher surface area (Khazaei et al., 2014; Jafari et al., 2007). Also, samples became with more smooth surfaces, rare cracks, and no dents which indicated a better protection of encapsulated oil and higher encapsulation efficiency (85.75%) which is in accordance with those results obtained by Assadpour and Jafari (2017).
Characterization of encapsulated powder by FTIR

FTIR method is nondestructive technique for studying the functional groups of food matrices including oil. The FTIR spectra of pomegranate seed oil (PSO) and encapsulated PSO with whey protein/ gum Arabic/ Maltodextrin are illustrated in Figure 3. In PSO, the peak located at 3009 cm⁻¹ corresponding to -C-H stretch which indicate to the double bond of an unsaturated fatty acid, while at 2925 and 2856 cm⁻¹ the peaks correspond to the alkene and alkane stretch vibrations of hydrocarbon. These peaks had stronger intensity in pure oil than encapsulated ones. The peak around 990-980 cm⁻¹ corresponds to the conjugated C-C double bonds (cis and trans) in punicic acid and their isomers. Peaks located at 3400 and 1637 cm⁻¹ in the FTIR spectra of encapsulated oil are attributed to the O-H stretching, and asymmetric N-H (-NH₃⁺) of amide I. The amide I peak at 1637 cm⁻¹ referred to the α-helix of secondary structure of proteins in whey protein (Dybing and Smith, 1991). Moreover, the main ingredient of whey protein is β-lactoglobulin (apex 57%) which provide emulsifying properties for their amphiphilic molecules that can increase emulsion stability and generate desirable characteristics of emulsions as well as decrease the interfacial tension between the hydrophilic and lipophilic phases (Sobhaninia et al., 2017). The peak at 1741 cm⁻¹ represents to the carbonyl stretching group of the triacylglycerols in unsaturated fatty acids.

Jelly form of encapsulated oil

The nano-encapsulation process is an alternative technique to increase the oil stability and allow its incorporation in foods with high water content. Addition of pomegranate seed oil in the form of nano-capsules to jelly was done to improve the bioavailability of the pomegranate seed oil. Jelly was selected as an application because it is a dietary supplement of a delicious taste and of high interest, especially for children. Figure 4 showed the jelly fortified with pomegranate seed oil nano capsules with different concentration of addition, comparing to control sample.
The obtained results lead to a good characteristic of encapsulated oil with water soluble properties and accordingly, more bioavailability. Finally, formation of nano-encapsulated oil in the jelly shapes made it easy and attractive to the consumer as a source of dietary supplement of punicic acid with reported good absorption and great bioavailability.

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